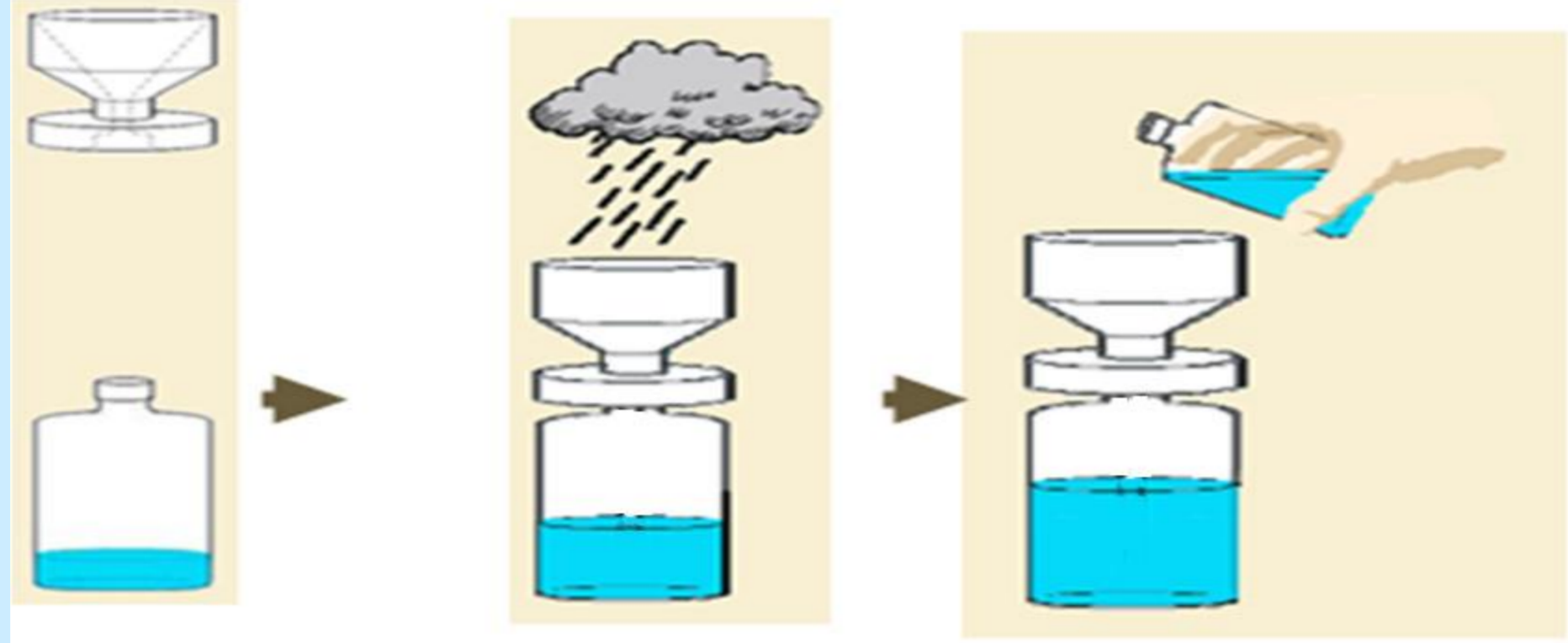


Introduction:

A lot of studies have shown the difficulty to collect and preserve trace metals in wet deposition samples. Windom and al. (1991) have shown that chemical contamination have compromised measurement in seawater for many years. Benoit and al. (1997) explained the same difficulties in freshwater. Chemical contaminations compromise measurements by increasing original concentrations. Also targeted metals may escape from the sample by absorption, evaporation or diffusion. Mercury is one of the most difficult metals to preserve because Hg is ubiquitous in ambient air as gaseous. The choice of the sampling flask (Amina et al. (1999)), the preservative reagent used, and the cleanup sampling material (Gasparon and al. (1998)) can influence the quality of analytical results. The behavior of mercury storage is completely different compared to other trace metals. Sampling in remote areas induce large delays before analysis (more than 6 month in our case) that considerably enhance contaminations. We present here methods applied to cleanup sampling materials for ultra-trace dissolved metals, including mercury, adapted for long-term storage.

Total atmospheric deposition metal trace method sampling used:

Total deposition is collected using Teflon PTFE machined funnels with a 12 cm diameter. A bottle containing storage solution is screwed at the bottom of the funnel. All this system stands 2 m high at the top of a 10 cm diameter PVC tube secured with Kevlar ropes. After sample collection, the system is rinsed with 60 ml of the storage solution. Next, the bottle is firmly closed and stored in a double sealed bag until analyse. All blank results shown here are experimental field blank.



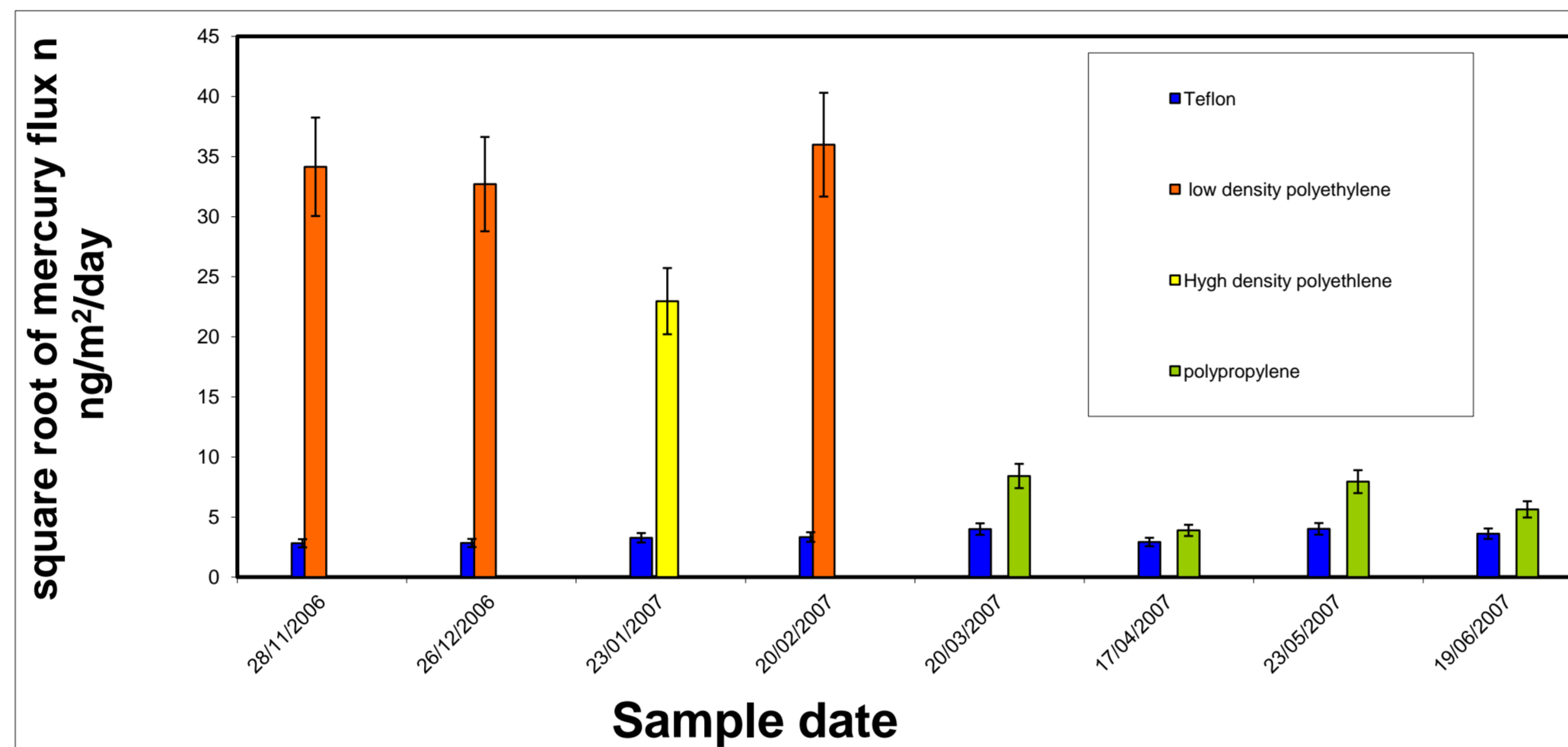
Sample collection protocol



Collector rinsing (Kerguelen island)

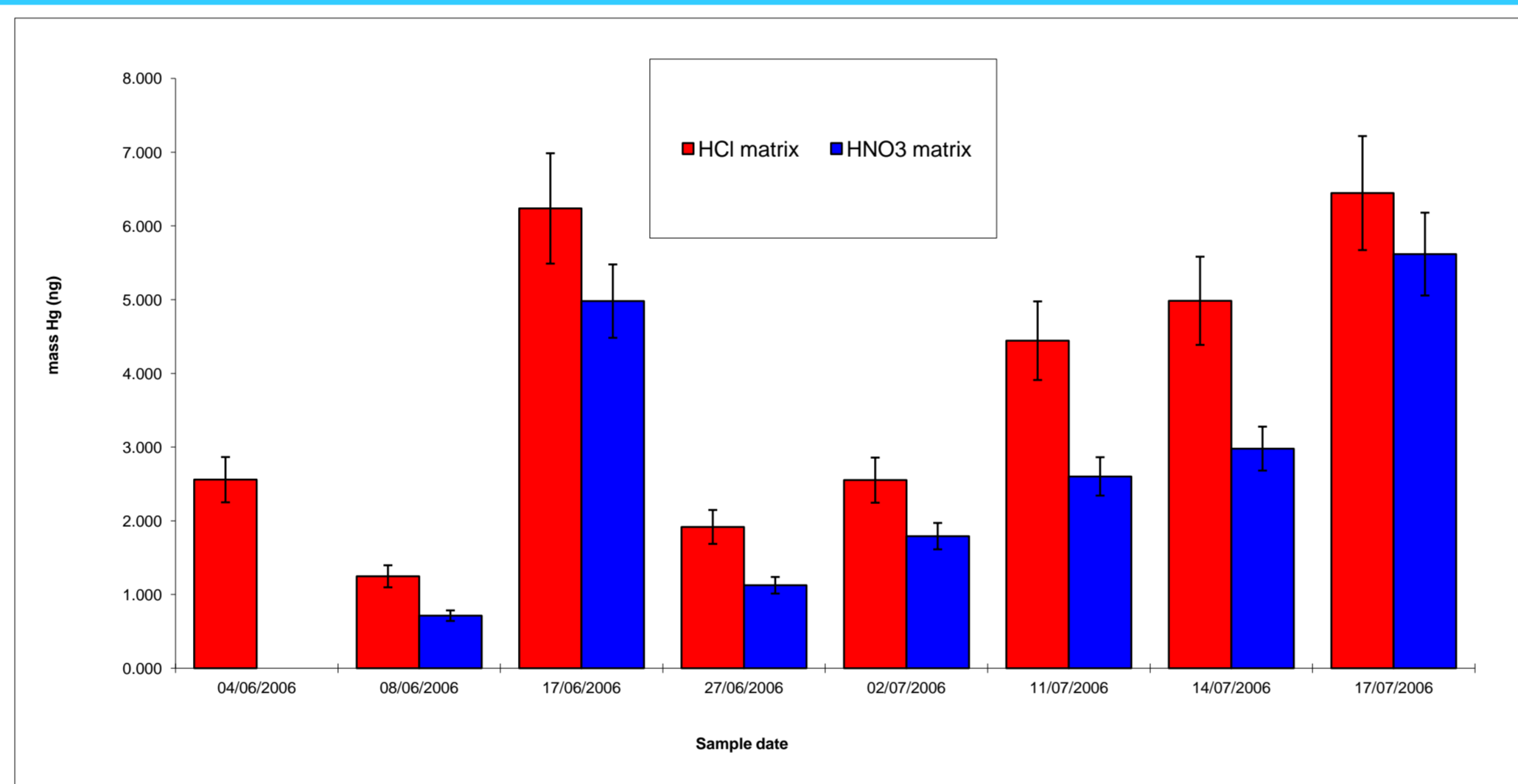
Mercury storage: Bottle material

For the choice of the bottle material, series of samples collected with each time FEP and an another material are compared. All samples are stored with ultra pure hydrochloric acid 2% v/v. **We can observe a good storage for Teflon bottles.**



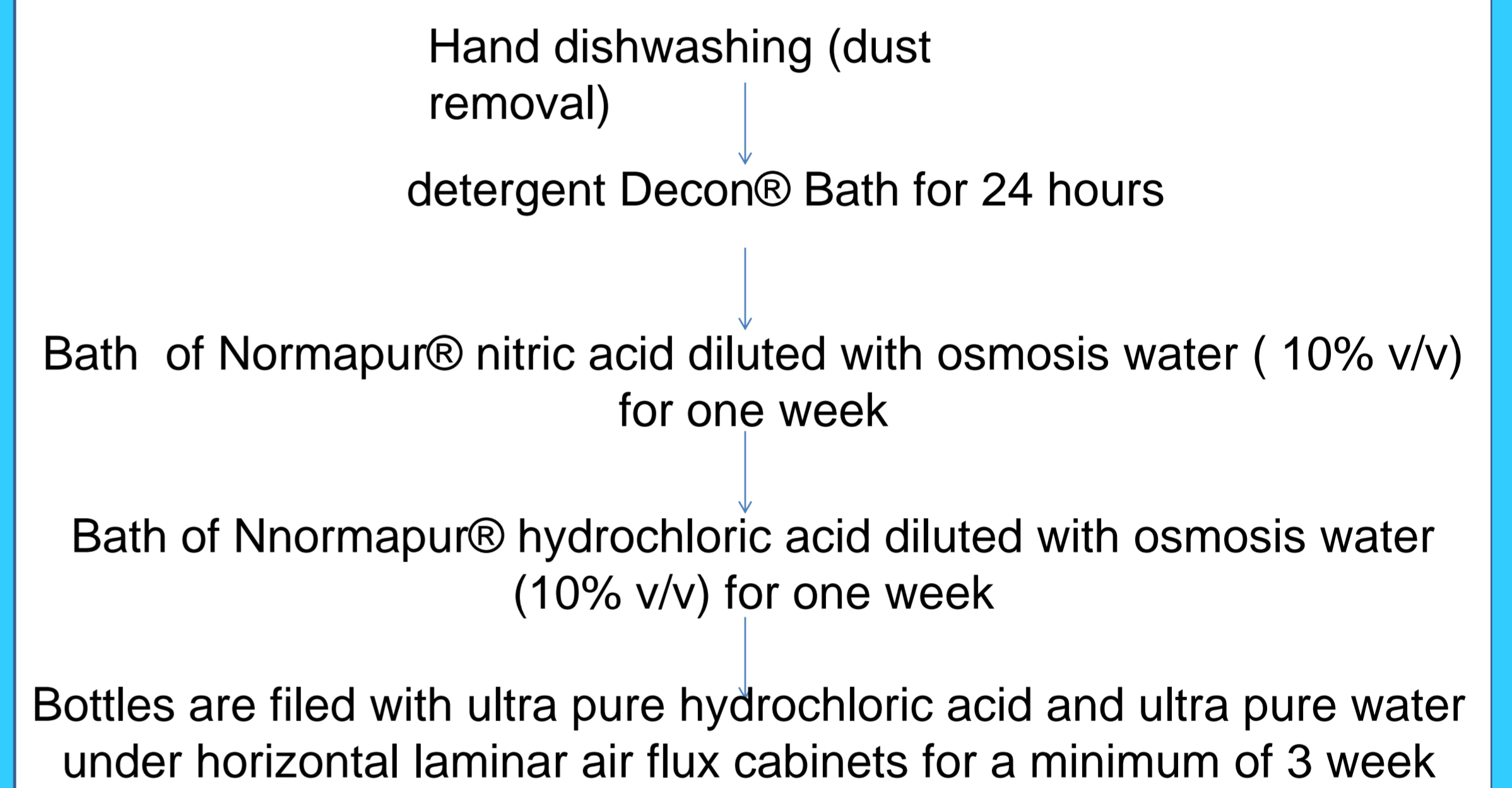
Mercury storage: acid used

To test the choice of acid, series of 2 collecting samples in Teflon bottles containing hydrochloric and nitric acid (2% v/v) were compared. **We can observe a better storage with hydrochloric acid.**



Procedure of bottles decontamination

The decontamination method was performed using Cossa et al. 2003 procedure and standard EPA 1631 revision E protocol:



Site de collecte	Mass of mercury in blanks (pg)	RSD	Approximate mass collected in each samples (pg)
Vouzon (France)	176	26%	1000
Niger (Africa)	214	No calculated	3000
Kerguelen (TAAF)	160	37%	550 for 5 samples

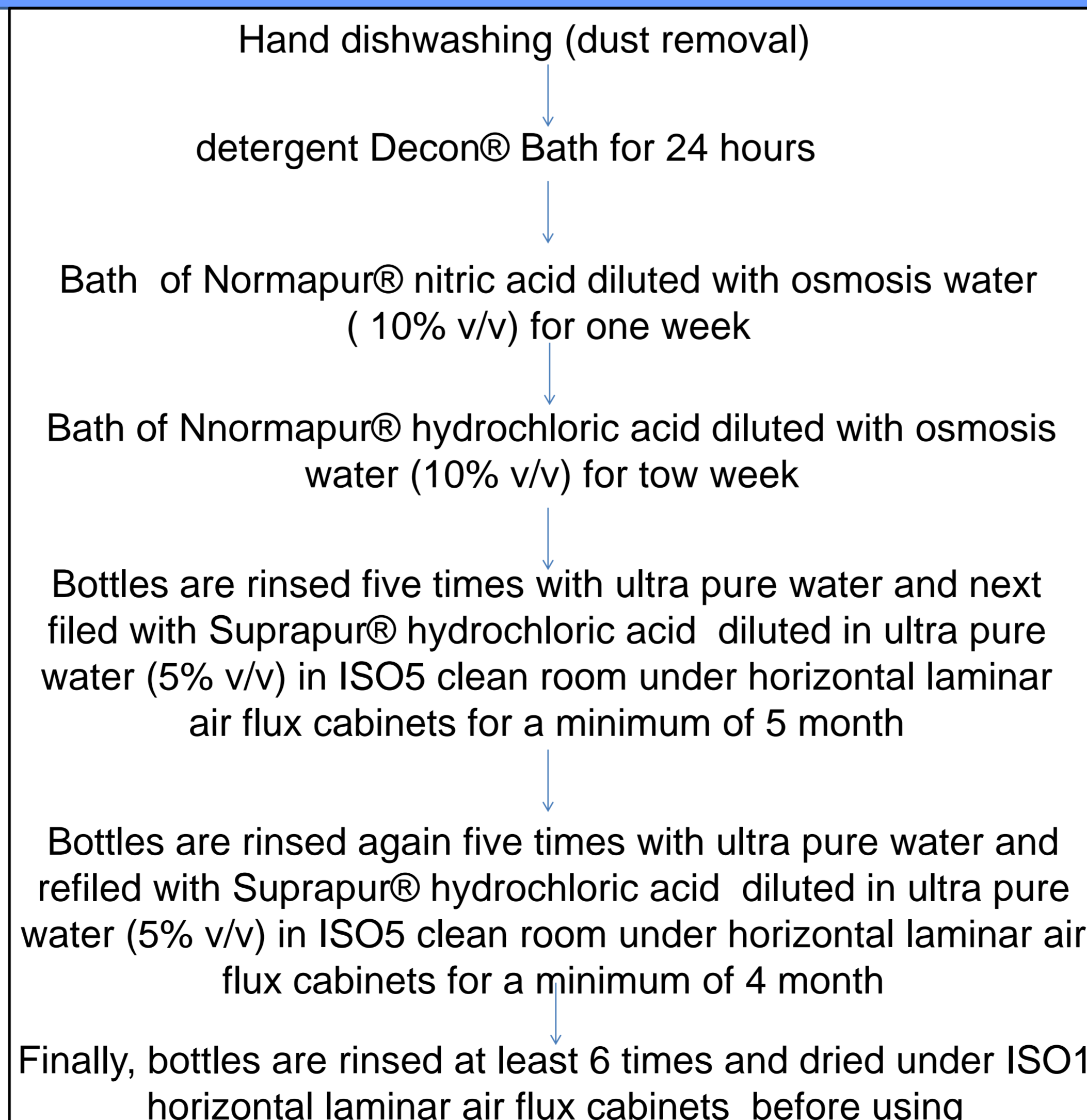
the storage method of environmental samples of mercury is effective enough for provides high quality measurements even with a long time between the collection and the analysis

Other trace metals storage:

Hydrochloric acid is generally used to clean bottles but **it's important to clean them with nitric acid to ensure the total removal of contamination from the surface of the bottle because ultra pure nitric acid (2% v/v) is then used to store the sample (HNO₃ matrix).**

For the same reason, the **concentration of acid used must be higher than the concentration of the matrix.**

For cost problem, polypropylene bottles are chosen instead of Teflon material. **PP can be used if the washing procedure is strictly executed.**



Analytical results from Kerguelen Island (Southern Ocean)

	Fe µg/L	Al µg/L
Analytical blanks	0.1	0.02
Field blanks	1.7 ± 0.35	3.9 ± 1.4

Expressed as amounts, filed blanks are never larger than 10% of the deposit quantities in any collected samples

we can say that the ultra clean washing method is validated